

*The Department of Mechanical Engineering presents:*

***The Master's Dissertation Defense of:***

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**One-Step Fabrication of Photocatalytic Nanoporous Titania for  
Efficient Water Treatment**

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Titanium dioxide has been shown to degrade organic contaminants in water through photocatalysis. While current research focuses on maximizing degradation efficiency, a need exists to further investigate the oxidation properties of titania as a means to tailor this material for integration into a high-throughput device. Herein we present a nanoporous titania (NPT) exhibiting good degradation efficiency as groundwork for a microfluidic reactor. Using hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) oxidation of titanium substrates, we were able to grow a high surface area nanoporous film with good structural integrity, and little to no signs of delamination. A growth parameter study was conducted to determine the optimal oxidation conditions for photocatalytic activity. Characterization of the material was carried out with XRD and SEM to record the effects of H<sub>2</sub>O<sub>2</sub> concentration, temperature, and time on the morphology and crystallinity of the NPT. In order to establish a standard, we used BET analysis to calculate a surface area of Degussa P25 comparable to our NPT 1x1 in<sup>2</sup> chip. Photocatalytic response was measured and compared to P25 drop-casted films via degradation of methylene blue. Over the course of three hours, we observed a 56.13% and a 76.16% degradation of methylene blue from our NPT and P25 samples, respectively. The degradation of methylene blue was found to be a first order reaction with the two materials demonstrating reaction rate constants of 0.0045 (NPT) and 0.0081 (P25) mg/L min<sup>-1</sup>. The studies conducted show that NPT is an efficient photocatalyst with good structural integrity to suit applications in microfluidic reactors.